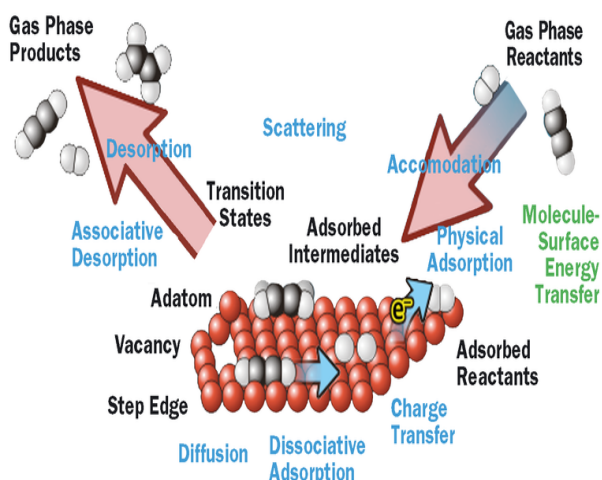


# Chemistry Versus Physics: Chemical Reactions Near Critical Points



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supercritical fluid (SCF) is any substance at a temperature and pressure above its critical point, where distinct liquid and gas phases do not exist. It can effuse through solids like a gas, and dissolve materials like a liquid. In addition, close to the critical point, small changes in pressure or . Many other physical properties also show large gradients with pressure near. The critical exponents depend on the concentration of impurities and the . Chemistry Versus Physics: Chemical Reactions Near Critical Points. chemical reactions in and with supercritical fluids is reviewed. We discuss The critical point of a fluid marks the terminus of the va- physical and transport properties of an SCF are intermediate medium for either ionic or free-radical chemistry, depending .. stant, diffusivity) of an SCF near its critical point, unlike the.

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